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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.041$
$w R$ factor $=0.114$
Data-to-parameter ratio $=12.2$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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$N$-(4-Methyl-1-piperazinyl)phthalimide

The title compound, $\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2}$, (I), crystallizes in the monoclinic space group $P 2_{1} / m$. The phthalimide rings lie in crystallographic mirror planes. Molecules are associated into columns parallel to the $b$-axis direction, and linked together along the $a$-axis direction by a two-dimensional network of hydrogen bonds involving $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions. The molecular packing in the crystal is stabilized by the weak intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and van der Waals forces.

(I)

## Experimental

The title compound was prepared by the reaction of phthalic anhydride ( 6.4 g ) with 1-amino-4-methylpiperazine ( 5.0 g ) under microwave irradiation for 10 min . The resulting product was heated and dissolved in ethanol ( 60 ml ). The homogeneous solution was allowed to stand at room temperature for 12 h , after which 7.0 g of the colorless crystalline product was separated by filtration. Pure $N$-(4-methyl-1-piperazinyl)phthalimide ( 1.5 g ) was dissolved in ethanol $(20 \mathrm{ml})$. A single crystal was obtained by evaporation for 10 h at room temperature.

## Crystal data

$\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2}$
$M_{r}=245.28$
Monoclinic, $P 2_{1} / m$
$a=8.364(3) \AA$
$b=6.816(3) \AA$
$c=10.972(5) \AA$
$\beta=101.98(6)^{\circ}$
$V=611.9(4) \AA^{\circ}$
$Z=2$
$D_{x}=1.331 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 853
reflections
$\theta=2.8-26.4^{\circ}$
$\mu=0.09 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Prism, colorless
$0.24 \times 0.22 \times 0.16 \mathrm{~mm}$

## Data collection

| Bruker SMART CCD area-detector | 1372 independent reflections |
| :--- | :--- |
| diffractometer | 1064 reflections with $I>2 \sigma(I)$ |
| $\varphi$ and $\omega$ scans | $R_{\text {int }}=0.021$ |
| Absorption correction: multi-scan | $\theta_{\max }=26.4^{\circ}$ |
| $(S A D A B S ;$ Sheldrick, 1996 $)$ | $h=-10 \rightarrow 10$ |
| $T_{\min }=0.976, T_{\max }=0.985$ | $k=-8 \rightarrow 8$ |
| 3540 measured reflections | $l=-8 \rightarrow 13$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.041$
$w R\left(F^{2}\right)=0.114$
$S=1.08$
1372 reflections
112 parameters
H atoms treated by a mixture of independent and constrained refinement

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Figure 1
The molecular structure of (I), showing $30 \%$ probability displacement ellipsoids. [Symmetry code: (i) $x, \frac{1}{2}-y, z$.]

Table 1
Hydrogen-bonding geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| C5-H5 $\mathrm{OO}^{\mathrm{O}} 1^{\mathrm{ii}}$ | 0.93 | 2.36 | $3.249(3)$ | 161 |
| C5-H5 $^{\mathrm{iii}}$ | 0.93 | 2.36 | $3.249(3)$ | 161 |

Symmetry code: (ii) $1+x, y, z$; (iii) $1+x, \frac{1}{2}-y, z$.

H atoms were treated as riding, with $\mathrm{C}-\mathrm{H}$ distances of $0.93-$ $0.97 \AA$. For the H atoms attached to atom $\mathrm{C} 13, U_{\text {iso }}(\mathrm{H})=$ $1.5 U_{\text {eq }}(\mathrm{C} 13)$; all other $U_{\text {iso }}(\mathrm{H})$ values were refined.

Data collection: SMART (Bruker, 1997); cell refinement: SMART; data reduction: SAINT (Bruker, 1997); program(s) used to solve structure: SHELXTL (Sheldrick, 2001); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.


Figure 2
Packing diagram showing $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions, viewed down the $b$ axis.

## References

Bruker (1997). SMART (Version 5.051) and SAINT (Version 5.A06). Bruker AXS Inc., Madison, Wisconsin, USA.
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